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Lesson 7.1 Reading and Study Workbook • Copyright © Pearson Education, Inc., or its affiliates. All Rights Reserved. 83. Ionic and Metallic Bonding. BONDING AND INTERACTIONS. 7.1 Ions. Essential Understanding Ions form when atoms gain or lose valence electrons, becoming electrically charged. Lesson Summary. Valence Electrons Valence electrons are the electrons in the outermost occupied energy level and are involved in ion formation. For a representative element, the group number equals the ...

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Chemical Bonding Types of chemical bonds Lewis structures and the octet rule Ionic and covalent bonding Bond polarity and electronegativity Formal charge Chapter 8 Energetics of bond formation and dissociation Bond formation: $A + B \rightarrow AB$ If the reaction is energetically favorable, AB is lower in energy than $A + B$. Heat is released .

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